

### MUTHAYAMMAL ENGINEERING COLLEGE

(An Autonomous Institution)

(Approved by AICTE, New Delhi, Accredited by NAAC & Affiliated to Anna University) Rasipuram - 637 408, Namakkal Dist., Tamil Nadu.



MKC

2020-21

### MUST KNOW CONCEPTS

### MECH

Course Code & Course Name :

### 16MEE03 & Advanced I.C Engines

Year/Sem/Sec

: IV/VII/B

S.No.	Term	Notation	Concept/Definition/Meaning/	Units
0.110.	Term	(Symbol)	Units/Equation/Expression	CIIIIS
		Un	it-I : Introduction	
1	Engine		The part of a vehicle that produces power to	
1.	Lingine		make the vehicle move.	
2.	Heat Engine		A heat engine is a device which transforms the	
	Types of Heat		(a) External combustion angine	
3.	Engine		(a) External combustion engine	
			In this engine, the products of combustion of air	
4	External		and fuel transfer heat to a second fluid which is	
т.	Combustion Engine	$\sim$	the working fluid of the cycle	
			In this engine, the combustion of air and fuels	
5	Internal Combustion		take place inside the cylinder and are used as the	
	Engine		direct motive force.	
		$\langle \rangle$	1. Cylinder	
	Components of Reciprocating IC		2. Piston	
6			3. Connecting rod	
0.	Engine	SIGNING.	YO 4.R Crankshaft E	
	Lingino	1	5. Crank case	
		Estd.	6. Flywheel	
	Truess of Easy Stualto		I. Suction stroke     Communication stroke	
7.	Types of Four Stroke		2. Compression stroke	
	Engine		J. Expansion shoke       A. Exhaust stroke	
0	Combustion		The process of huming	
0.	Combustion		The process of burning.	
9.	Types of Combustion		a) Normal Combustion	
			b) Admonthal Compusition	
	Normal and		of the compression stroke at the spark plug by an	
10	Abnormal		electric discharge	
10.	Combustion		<b>Abnormal combustion</b> in spark ignition engine	
			majorly occurs as knock and surface ignition.	
			Cylindrical component fitted into cylinder	
11.	Piston		forming the moving boundary of the combustion	
			system.	

12.	Connecting Rod		Interconnects the piston and crankshaft.	
13.	Crankshaft		Major engine component which converts the reciprocating motion of the piston into rotary motion.	
14.	Camshaft		Receives the drive from crankshaft and control the valve opening.	
15.	Valve		To admit the air-fuel mixture in engine cylinder and discharging the products of combustion from cylinder.	
16.	Carburetor		Atomizes the fuel and mixes it with air.	L
17.	Unit Injector		Combination of high pressure pumps and injectors in one unit.	
18.	Ignition system		Produce spark in injection cylinder towards the end of the compression stroke.	
19.	Smog		Mixture of particles of unburnt fuel and the air.	
20.	Knock		Knock is the auto ignition of the portion of Fuel, air and residual gas mixture ahead of the advancing flame that produces noise.	
21.	Surface Ignition	X	Surface ignition is ignition of the fuel air charge by overheated valves or spark plugs.	
22.	Catalytic Converter	$\sim$	Harmful gases converted into Harmless gases.	
23.	Effect of engine variables on flame propagation	$\gtrsim$	<ol> <li>Air fuel ratio</li> <li>Compression ratio</li> <li>Load on engine</li> <li>Turbulence and engine speed</li> <li>Other factors</li> </ol>	
24.	Detonation		Combustion of a substance which is initiated suddenly and propagates extremely rapidly, giving rise to a shock wave.	
25.	Effect of engine operating variables on the engine knocking	Estd.	<ul> <li>a) Temperature Factor</li> <li>b) Density Factor</li> <li>c) Time Factor</li> <li>d) Composition Factor</li> </ul>	
	Unit-II	: COMPRE	ESSION IGNITION ENGINES	
26.	Types of diesel injection system		<ul> <li>a) Air injection system</li> <li>b) Solid injection system</li> <li>c) jerk pump system</li> <li>d) Common rail system</li> <li>e) Distributor system</li> </ul>	
27.	Blast injector		These are superseded by mechanically operated injectors used in air injection system	
28.	Multi-hole nozzle		Number of hole varies from 4 to 18 and the size from 1.5 to 0.35 mm and the injection rate is not uniform	
29.	Single hole nozzle		It is used in open combustion chamber. The size of hole larger than 0.2 mm and very high injection pressure is required	

30.	Turbocharger		By utilizing the exhaust energy to drive the gas turbine.	
31.	Supercharger		Increase the air density for maximize the power output	
32.	Types of Combustion Chamber		<ol> <li>Direct Injection (or) open Injection Type</li> <li>Indirect Injection Type</li> </ol>	
33.	Types of open combustion chamber		<ol> <li>Shallow Depth Chamber</li> <li>Hemispherical Chamber</li> <li>Cylindrical Chamber</li> <li>Toroidal Chamber</li> </ol>	
34.	Important fuel specifications for diesel		<ul><li>a) Viscosity</li><li>b) Surface tension</li><li>c) Cetane number</li></ul>	
35.	Surface Tension		The Surface tension is a parameter which effects the formation of fuel droplets in sprays.	
36.	Cetane number		50 - 60 for high speed Diesel engines 25 - 45 for low speed Diesel engines Normal diesel fuel CN is 40-50	
37.	Atomization		It refers to separating something into fine particles. It is a process of breaking bulk liquids into small droplets.	
38.	Stages of combustion in C.I Engine	$\langle \cdot \rangle$	Stage 1: Ignition delay period Stage 2: Period of rapid combustion. Stage 3: Period of controlled combustion. Stage 4: Period of after burning.	
39.	Factors Affecting the delay period	$\mathbf{X}$	<ol> <li>Compression ratio</li> <li>Atomization of the fuel</li> <li>Quality of the fuel</li> <li>Intake temperature and pressure</li> </ol>	
40.	Applications of swirl chamber	X	<ul><li>a. Where fuel quality is difficult to control</li><li>b. Where reliability under adverse condition is</li><li>more important than fuel economy</li></ul>	
41.	Parts of Turbocharger	Estd.	<ul> <li>a) Turbine Wheel,</li> <li>b) Turbine Housing,</li> <li>c) Turbo Shaft,</li> <li>d) Compressor Wheel,</li> <li>e) Compressor Housing And</li> <li>f) Bearing Housing.</li> </ul>	
42.	Turbo Lag		It refers to the short delay period before the boost or manifold pressure increase .	
43.	Flame development angle		The crank angle interval between the spark discharge and the time when a small but significant fraction of the cylinder mass has burned or fuel chemical energy has been released	
44.	Rapid burning angle		The crank angle interval required to burn the bulk of the charge is defined as the interval between the end of the flame development stage and the end of the flame propagation process.	
45.	Parameters of Macroscopic Characteristics		<ul><li>1.Spray tip penetration</li><li>2.Spray angle</li><li>3.Break up length</li></ul>	

46.	Factors Influence Spray Penetration in CI Engines		<ul> <li>a) Spray Formation</li> <li>b) Spray Characteristics</li> <li>c) Spray Penetration</li> <li>d) Spray Direction</li> </ul>
47.	Oxygen concentration		Residual gases reduce O <sub>2</sub> concentration and reducing oxygen concentration increases ID.
48.	Surface Ignition		Surface ignition is ignition of the fuel air charge by overheated valves or spark plugs.
49.	Knock		Knock is the auto ignition of the portion of Fuel, air and residual gas mixture ahead of the advancing flame that produces noise.
50.	Types of Combustion		<ul><li>c) Normal Combustion</li><li>d) Abnormal Combustion</li></ul>
	Unit-III : P	OLLUTAN	T FORMATION AND CONTROL
51.	Exhaust Emissions		Internal combustion engines operate by burning of the fossil fuel derivatives. The exhaust emissions are the major contribution to environmental pollution.
52.	Major exhaust emissions	$\mathbf{X}$	<ol> <li>Unburnt gas</li> <li>Oxides of carbon (co and co2)</li> <li>Oxides of nitrogen (NO and NO2)</li> <li>Oxides of Sulphur (SO2 and SO3)</li> </ol>
53.	Mechanism of NO formation	$\otimes$	The nitric oxide formation during the combustion process is the result of group of elementary reactions involving the nitrogen and oxygen.
54.	Simple reaction (N2 O2)	$\langle \cdot \rangle$	N2 + O2→2NO
55.	Zelodovich chai reaction mechanism		O2→2O O+N2→NO+N N+O2→NO+O
56.	Hydrocarbons formations	Eatd	Wall quenching, Incomplete combustion of fuel, Exhaust scavenging in 2-stroke engines.
57.	Wall quenching	<u>cstu</u> .	The quenching of the flame near the combustion chamber walls is known as the wall quenching.
58.	Incomplete combustion		Under the operating conditions, where mixtures are extremely rich or lean or exhaust gas dilution is excessive, incomplete flame propagation's occurs during the combustion is called incomplete combustions
59.	Factors of Incomplete combustion		<ul><li>1.Poor condition of ignition system including spark plug</li><li>2.Low charge temperature</li><li>3.Too rich or lean mixture in cylinder</li></ul>
60.	Scavenging		In 2-stroke engine a third source of HC emission results from scavenging of the cylinder with fuel air mixture.
61.	Carbon monoxide's formation		Carbon monoxide remains in the exhaust if the oxidation of the CO and CO2 is not complete it

62.       Diesel engine smoke emission        Engine exhaust smoke is a visible indicator of the combustion process in the engine. Smoke is due to incomplete combustion.         63.       Types of diesel engine smoke        1. blue smoke         64.       Blue smoke        2. white or cold smoke         65.       White or cold smoke        it results from the burning of engine lubricating oil that reaches combustion chamber due the worm piston rings, cylinder liners and valve guides         65.       White or cold smoke        name up of the droplets of unburnt or partially burnt fuel droplets of suburnt or partially burnt fuel droplets of unburnt or partially burnt fuel droplets of unburnt or partially burnt fuel droplets of suburnt or partially burnt fuel droplets of suburnt or partially burnt fuel droplets of unburnt or partially and other solid products of combustion.         66.       Black or hot smoke        It is consisting of unburnt carbon particles (0.5"1 micron in diameter) and other solid products of combustion.         67.       Particulate's        Particulate's matter comes from the hydrocarbons, lead additives and Sulphur dioxide. Very harmful to humans, animals, plants, and nature.         68.       formations        The greenhouse ensults and lead compounds resulting from the use of TEL are exhaust gas from the CLSI engines. Size of particles (0.02 to 0.06). <th></th> <th></th> <th></th> <th>is an intermediate part of the combustion</th>				is an intermediate part of the combustion
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62.       Description Sinoke emission        the combustion process in the engine. Smoke is due to incomplete combustion.         63.       Types of diesel        1.       blue smoke          64.       Blue smoke        2.       white or cold smoke          64.       Blue smoke        0.       It results from the burning of engine lubricating oil that reaches combustion chamber due the worn piston rings, cylinder liners and valve guides.         65.       White or cold smoke        associated with the engine running at the less than the normal operating temperature after starting.         66.       Black or hot smoke        It is consisting of unburnt carbon particles (0.5°1 micron in diameter) and other solid products of combustion.         67.       Particulate's        Particulate's matter comes from the hydrocarbons, lead additives and Sulphur dioxide. Very harmful to humans, animals, plants, and nature.         68.       Particulate's formations        Organic and inorganic compounds of higher molecular weights and lead compounds resulting from the use of TEL are exhaust gas from the CLSI engines. Size of particles (0.02 to 0.06).         69.       Greenhouse gases        Acatalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         71.       Catalysts        1. P		Diasal angina smoka		Engine exhaust smoke is a visible indicator of
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69.       Greenhouse effects        The greenhouse effect is a process by which thermal radiation from a planetary surface is absorbed by atmosphere greenhouse re-radiation in all directions.         70.       Greenhouse gases        Water vapor 36-70% Carbon dioxide 9-26% Methane 4-9% Ozone 3-7% Ozone 3-7%         71.       Catalysts        A catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         72.       Material used in catalyst        1. Platinum         73.       Methods of measuring emissions        1. Oxides of nitrogen         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.       Inti-IV :: ALTERNATIVE FUELS	66.	formations		from the use of TEL are exhaust gas from the
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69.       Greenhouse effects        thermal radiation from a planetary surface is absorbed by atmosphere greenhouse re-radiation in all directions.         70.       Greenhouse gases       Greenhouse gases       Water vapor 36-70% Carbon dioxide 9-26% Methane 4-9% Ozone 3-7%         71.       Catalysts        A catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         72.       Material used in catalyst        1. Platinum         73.       Methods of measuring emissions        2. Carbon monoxide         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.       Invisible emissions				The greenhouse effect is a process by which
09.       Oreenhouse enects       absorbed by atmosphere greenhouse re-radiation in all directions.         70.       Greenhouse gases       Water vapor 36-70% Carbon dioxide 9-26% Methane 4-9% Ozone 3-7%         71.       Catalysts       A catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         72.       Material used in catalyst        1. Platinum         73.       Methods of measuring emissions        2. Carbon monoxide         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.         Unit-IV: : ALTERNATIVE FUELS	60	Graanhousa offacts		thermal radiation from a planetary surface is
70.       Greenhouse gases       IGNINC       Carbon dioxide 9-26% Methane 4-9% Ozone 3-7%         71.       Catalysts        A catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         72.       Material used in catalyst        1. Platinum 2. Palladium 3. Rhodium         73.       Methods of measuring emissions        2. Carbon monoxide 3. Unburned hydrocarbons         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.         Unit-IV :: ALTERNATIVE FUELS	09.	Greenhouse effects		absorbed by atmosphere greenhouse re-radiation
70.Greenhouse gasesIGNING IGNING Carbon dioxide 9-26% Methane 4-9% Ozone 3-7%71.Catalysts71.Catalysts72.Material used in catalyst72.Material used in catalyst73.Methods of measuring emissions74.Invisible emissions75.Visible emissions75.Visible emissions75.Visible emissions76.Unit-IV :: ELTERNATIVE FUELS				in all directions.
To.       Greenhouse gases       Gasting of the sector of the sec				Water vapor 36-70%
Methane 4-9% Ozone 3-7%         Methane 4-9% Ozone 3-7%         A catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.         Material used in catalyst        1. Platinum         Palladium       3. Rhodium         Methods of measuring emissions        2. Carbon monoxide         Number of the emissions        2. Carbon monoxide         Mater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons       Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         Visible emissions        Smoke, particulate.       Smoke, particulate.	70	Greenhouse gases	<u>IGNING</u>	Carbon dioxide 9-26%
71.CatalystsA catalyst is a substance that accelerates chemical reaction by lowering the energy needed for it to proceed.72.Material used in catalyst1. Platinum 2. Palladium 3. Rhodium73.Methods of measuring emissions1. Oxides of nitrogen 2. Carbon monoxide 3. Unburned hydrocarbons74.Invisible emissionsWater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons75.Visible emissionsSmoke, particulate.Unit-IV :: ALTERNATIVE FUELS	70.	Orcennouse gases		Methane 4-9%
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71.       Catalysts        chemical reaction by lowering the energy needed for it to proceed.         72.       Material used in catalyst        1. Platinum         72.       Material used in catalyst        2. Palladium         73.       Methods of measuring emissions        1. Oxides of nitrogen         73.       Methods of measuring emissions        2. Carbon monoxide         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.         Unit-IV :: ALTERNATIVE FUELS				A catalyst is a substance that accelerates
Image: constraint of the proceed.Image: constraint of the proceed.72.Material used in catalyst1.Platinum73.Methods of measuring emissions2.Palladium73.Methods of measuring emissions2.Carbon monoxide74.Invisible emissionsWater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons75.Visible emissionsSmoke, particulate.Unit-IV :: ALTERNATIVE FUELS	71.	Catalysts		chemical reaction by lowering the energy needed
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72.Interfact about in catalyst2. Palladium 3. Rhodium73.Methods of measuring emissions1. Oxides of nitrogen 2. Carbon monoxide 3. Unburned hydrocarbons74.Invisible emissionsWater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons75.Visible emissionsSmoke, particulate.Unit-IV :: ALTERNATIVE FUELS		Material used in		1. Platinum
Catalyst3. Rhodium73.Methods of measuring emissions1. Oxides of nitrogen 2. Carbon monoxide 3. Unburned hydrocarbons74.Invisible emissionsWater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons75.Visible emissionsSmoke, particulate.Unit-IV :: ALTERNATIVE FUELS	72.	catalyst		2. Palladium
73.Methods of measuring emissions1. Oxides of nitrogen 2. Carbon monoxide 3. Unburned hydrocarbons74.Invisible emissionsWater vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons75.Visible emissionsSmoke, particulate.Unit-IV :: HITERNATIVE FUELS		cataryst		3. Rhodium
73.       Interform on measuring emissions        2. Carbon monoxide         74.       Invisible emissions        Water vapor, carbon dioxide, oxides of nitrogen, unburnt hydrocarbons         75.       Visible emissions        Smoke, particulate.         Unit-IV :: ALTERNATIVE FUELS		Methods of		1. Oxides of nitrogen
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74.     Invision classions     unburnt hydrocarbons       75.     Visible emissions        Smoke, particulate.       Unit-IV::ALTERNATIVE FUELS	74	Invisible emissions		Water vapor, carbon dioxide, oxides of nitrogen,
75.       Visible emissions        Smoke, particulate.         Unit-IV :: ALTERNATIVE FUELS	/ 1.			unburnt hydrocarbons
Unit-IV : : ALTERNATIVE FUELS	75.	Visible emissions		Smoke, particulate.
			Unit-IV : : A	LTERNATIVE FUELS
Non-petroleum based diesel fuel consists of		Die Diess1		Non-petroleum based diesel fuel consists of
76. BIO-Diesel mono alkyl esters derived from vegetable oil and	76.	B10-Diesel		mono alkyl esters derived from vegetable oil and
5				5

			animal fats.	
77.	Bio-gas		Gaseous fuel of varying proportions of methane, co <sub>2</sub> , water vapour etc	
78.	B-100		100% Bio-Diesel	
79.	Gasohal		90% gasoline with 10% anhydrous ethanol.	
80.	ethanol mixture		5 % anhydrous ethanol and 15 % gasoline.	
81.	Hybrid vehicle		Using two or more distinct power sources.	
82.	Fuel cell		Produces electricity through a chemical reaction, but without combustion.	
83.	Fuel cell parts		Anode, Cathode, Electrolyte and Fuel.	
84.	LPG		Liquified Petroleum Gas	
85.	CNG		Compressed Natural Gas, methane stored at high pressure.	
86.	Sources of methanol	<	coal, petroleum, natural gas, biomass, wood landfills and even theocean.	
87.	Sources of ethanol		Sugarcane, sugarbeets, and even cellulose (wood and paper).	
88.	Techniques of using alcohol	<	Alcohol dieselemulsions, Dual fuel injection, Alcohol fumigation, Surface ignition of alcohols.	
89.	Advantages of using hydrogen	$\langle \cdot \rangle$	Lowemissions, Fuelavailability, Fuel leakage to environment is not apollutant High energy continent per volume when stored as aliquid.	
90.	Disadvantages of using hydrogen	$\geq$	Difficult to re fuel, Fuel cost would be high at present day's technology and availability, Poor engine volumetric efficiency, High NOx emission because of highflame.	
91.	Methods for hydrogen usage in IC engine	IGNING	By manifoldinduction, By direct introduction of hydrogen into thecylinder, By supplementinggasoline.	
92.	Types of LPG	Estd.	One is propane and the other is butane	
93.	Advantages of LPG		LPG mixes with air at all temperatures. LPG has high antiknock characteristics. There is no crack case dilution, because the fuel is in the form of vapor.	
94.	Disadvantages of LPG		A special fuel feed system is required for liquid petroleum gas. A good cooling system is quite necessary. The vehicle weight is increased due to the use of heavy pressure cylinder for storing LPG	
95.	Piston		Cylindrical component fitted into cylinder forming the moving boundary of the combustion system.	
96.	Connecting rod		Interconnects the piston and crankshaft.	
97.	Crankshaft		Major engine component which converts the reciprocating motion of the piston into rotary	

			motion.
98.	Camshaft		Receives the drive from crankshaft and control the valve opening.
99.	Valve		To admit the air-fuel mixture in engine cylinder and discharging the products of combustion from cylinder.
100.	Sources of Methanol		Sugarcane, sugarbeets, and even cellulose (wood and paper).
		Unit-V :	RECENT TRENDS
101.	Carburetor		Atomizes the fuel and mixes it with air.
102.	Ignition system		Produce spark in injection cylinder towards the end of the compression stroke.
103.	Unit Injector		Combination of high pressure pumps and injectors in one unit.
104.	Supercharger		Increase the air density for maximize the power output
105.	Turbocharger		By utilizing the exhaust energy to drive the gas turbine.
106.	Catalytic Converter		Harmful gases converted into Harmless gases.
107.	EURO NORMS	X	Permissible emission levels which have been implemented in Europe.
108.	Smog		Mixture of particles of unburnt fuel and the air.
109.	Clutch		Connect or disconnect the power transmission.
110.	Gear box		Regulate both the power output and the speed range.
111.	Tractive effort	$\langle \cdot \rangle$	Driving force at driving wheel to propel the vehicle.
112.	Fluid flywheel		Hydraulic unit that replaces a clutch and transmits the engine torque to transmission system.
113.	Hotchkiss drive	Eatd	Open propeller shaft, in which the torque reaction is taken by the springs.
114.	Differential	LS <u>I</u> U.	Drives the outer wheel faster than the inner wheel while in turn.
115.	Live axle		Turns within a tubular housing.
116.	Dead axle		Solid axle mounted on springs with a spindle at each end.
117.	Power Steering		Operating the steering by using the compressed air or hydraulic pressure.
118.	Braking system		Reduce the speed, stop the moving vehicle and to hold the vehicle.
119.	Independent suspension		Mounting of the wheel on a separate axle.
120.	Wishbone		Triangular steel frame which connects vehicle body to each wheel in independent suspension system.
121.	Antilock Braking		Relieving the brake pressure momentarily to

	System		prevent locking of wheel.	
122.	Volatility		Evaporating tendency of a liquid fuel.	
123.	Flame speed		The speed at which flame travels inside the combustion chamber.	
124.	Pumping element		moves the fuel from the fuel tank to the injector. This include necessary piping, filter etc.	
125.	Timing control		fixes the start and stop of the fuel-air mixing process.	
		Plac	cement Questions	
126.	How many times are the hands of a clock at right angle in a day?		A. 22 B. 24 C. 44 D. 48 Explanation: In 12 hours, they are at right angles 22 times.	
127.	A train moves with a speed of 108 kmph. Its speed in metres per second is :		<ul> <li>In 24 hours, they are at right angles 44 times.</li> <li>A.10.8</li> <li>B.18</li> <li>C.30</li> <li>D.38.8</li> <li>Explanation:108 kmph = 108*[5/18] m/sec = 30</li> <li>m/s</li> </ul>	
128.	Determine the probability that a digit chosen at random from the digits 1, 2, 3,12 will be odd.	$\otimes$	Total no. of Digits = 12. Equally likely cases = 12. There are six odd digits. Probability = $6/12 = 1/2$	
129.	In covering a distance of 40 km, Kamlesh takes 2 hours more than Pankaj. If Kamlesh doubles his speed, then he would take 1 hour less than Pankaj. Then what is Kamlesh's speed?	Estd.	A. 11 kmph <b>B. 5 kmph</b> C. 9 kmph D. 6 kmph TURE Answer:B Explanation: Let Kamlesh's speed be x km/hr. Then, $40/x - 40/(2x) = 4$ 8x = 40 x = 5 km/hr	
130.	Solve the equation x+34=82		A. 58 B. 48 C. 55 D. 60 Explanation: x=82-34=48	
131.	An accurate clock shows 8 o'clock in the morning. Through how may degrees will the hour hand rotate when the clock		A.360. <b>B.180</b> C.90 D.60 Answer: B) 180 Explanation: Angle traced by the hour hand in 6	

	shows 2 o'clock in	hours=(360/12)*6	
	the afternoon?		
	Excluding stoppages,	A. 9	
	the speed of a bus is	<b>B.</b> 10	
132. 133. 133. 134. 135. 135. 136. 137. 138.	54 kmph and	C. 12	
	including stoppages,	D. 20	
132.	it is 45 kmph. For	Explanation:	
	how many minutes	Due to stoppages, it covers 9 km less.	
	does the bus stop per	Time taken to cover 9 km 9 x $_{min} = 10$	
	hour?	= 5460 min.	
	Find the no., when 15		
	is subtracted from 7	Let the analysis here	
133.	times the no., the	Let the number be $x$ .	
	result is 10 more than	7x - 15 = 2x + 10 => 5x = 25 => x = 5	
	twice of the number		
		A.1.12	
		B.1.16	
124	If 0.75: x :: 5:8, then	C.1.20	
134.	x is equal to:	D.1.30	
		Explanation: $(x * 5) = (0.75 * 8)$	
		X=6/5=1.20	
		A. Tuesday	
		B. Monday	
	Today is Monday	C. Sunday	
135.	After 61 days it will	D. Saturday	
155.	he ·	Answer: D) Saturday	
		Explanation: Each day of the week is repeated	
		after 7 days. So, after 63 days, it will be Monday.	
		After 61 days, it will be Saturday.	
	Adam can do a job in		
	15 days; Eve can do	Adam can do 1/15 of the job per day	
	the same job in 20	Eve can do 1/20 of the job per day	
136.	days. If they work	If they work together they can do 7/60 of the	
	together for 4 days, E	WG work together URE	
	what fraction of job	Remaining job 1 - $7/60 = 32/60 = 8/15$	
	is incomplete?	1.12000	
		A.31	
	Which one of the	B. 61	
137.	following is not a	C.71	
	prime number?		
	1	Explanation:	
		91 is divisible by 7. So, it is not a prime number.	
		B. 9	
100	<b>F</b> : 1 :65 2 22		
138.	rind c, ii 3c - 2 = 33	D. 15 Evaluation	
		Explanation:	
		we add 2 to both sides and get $5c-2+2=33+2$ , or 5a-25 We divide both sides by 5 to set $-7$	
		3c=35. we divide both sides by 5 to get $c=/$ .	
120	A person crosses a	A. 3.0 <b>P. 7.2</b>	
139.	5 minutes What is		
	5 minutes. what is	U. 8.4	

	his speed in km per		D. 10	
	nour :		Explanation:	
			Speed = $600/5 \times 60 \text{ m/sec.} = 2 \text{ m/sec.}$	
	A and B can do a		$= 2 \times 18/5$ km/hr $= 7.2$ km/hr	
	piece of work in 4			
	days, while C and D		A B C and D will together take $\frac{1}{4} + \frac{1}{12} - \frac{1}{12}$	
140.	can do the same work		= 1/3.	
	in 12 days. In how many days will A B		3 days to complete the work.	
	C and D do it			
	together?			
			A.25	
	The average of five		<b>B.35</b> C 45	
	numbers is 27. If one		D.55	
141.	the average becomes	-	Answer:B	
	25. The excluded		Explanation:	
	number is?		(2/*5)-(25*4) 135_100	
			35	
			A.4	
			<b>B.8</b>	
	The maximum gap		C.2	
142.	successive leap year	$\sim$	Answer: B) 8	
	is?	$\sim$	Explanation: This can be illustrated with an	
			example. Ex: 1896 is a leap year. The next leap	
		$\bigwedge$	year comes in 1904 (1900 is not a leap year).	
	A guy bought 10		A. 10% B. 5%	
	pencils for Rs. 50		C. 20%	
143.	and sold them for Rs.	IGNING	D.12% FUTURE	
	in terms of		Answer:C	
	percentage?	Estd.	Explanation: `"Goip%"-("Goip"/"C P")*100-20%`	
	Two trains starting at		- Gam 7 C.F ) 100–2070	
	the same time from 2			
	stations 200 km apart		In the same time, they cover 110 km and 90 km	
	and going in opposite		respectively.	
144.	other at a distance of		For the same time, speed and distance is	
	110 km from one of		inversely proportional.	
	the stations. What is		So ratio of their speed = $110.90 = 11.9$	
	the ratio of their			
	speeds? In 100 m race. A		A. 20m	
	covers the distance in		B. 25m	
145.	36 seconds and B in		C. 22.5m	
	45 seconds. In this		D. 9m	
	race A beats B by:		Explanation:	

			Distance covered by B in 9 sec. = $(100/45)$ *9m =	
			20m	
			A.0.2	
			B.0.02	
			C.0.005	
			D.0.05	
146.	Half percent, written		Answer: C	
	as a decimal, is			
			Explanation:	
			As we know, $1\% = 1/100$	
			Hence, $(1/2)\% = (1/2 * 1/100) = 1/200 = 0.005$	
	A nump con fill o		A. 4 1/3 Hours	
	A pump can mi a		B. 7 Hours	
	tank with water in 2		C. 8 Hours	
147	look it took 2.5 hours		D. 10 Hours	
147.	to fill the tenk. The		Explanation:	
	look con droin all the		Work done by the look in 1 hour $-(12)^{-1}$	
	water of the tenk in:		work done by the leak in 1 hour $-(2^{-5})^{-10}$ .	
	water of the talk m.	$\leq$	·· Leak will empty the tank in 10 hrs.	
	If a number is chosen			
	at random from 1 to		We have 1,8,27 and 64 as perfect cubes from 1	
140	100, then the		to 100.	
140.	probability that the		Thus, the probability of picking a perfect cube is	
	chosen number is a		4/100 = 1/25	
	perfect cube is			
			A. 9	
	Three times the first		B. 11	
	of three consecutive	<	C. 13	
140	odd integers is 3		D. 15	
149.	more than twice the		Explanation:	
	third. The third	$< \land$	Let the three integers be $x$ , $x + 2$ and $x + 4$ .	
	integer is:		Then, $3x = 2(x + 4) + 3 \iff x = 11$ .	
			$\therefore$ Third integer = $x + 4 = 15$ .	
	Eind the number DE	SIGNING	A.5JR FUTURE	
	Find the number,		B. 15	
	from 7 times the	Fetd	C. 7.5	
150.	number the regult is	LJLU.	D. 4	
	10 more then twice		Explanation:	
	of the number		Let the number be x.	
	or the number		$7x - 15 = 2x + 10 \Longrightarrow 5x = 25 \Longrightarrow x = 5$	

## **Faculty Team Prepared**

# Signatures

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- Mr.M.Soundarrajan.
   Mr.M.Maniyarasan

